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Solution: Let a be the length of the sides of the tetrahedral cube and r be the radius of the two types of atoms.

$$a = 2r_C + 2r_H$$

$$a = 2r_C + 2r_H = 2$$

$$(0.77 \times 10^{-8} \text{ cm} + 0.46$$

$$\times 10^{-8} \text{ cm}) = 1.42$$

$$\times 10^{-8} \text{ cm} = 1.42 \times 10^{-8} \text{ cm}$$

37.2–20 The

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phosphide (AlP) is a compound semiconductor having mixed ionic and covalent bonding.

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2-10 Write the electronic configuration of the following elements (a)...

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