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STRUCTURE OF ATOM 27 ••• to explain the formation of different kinds of molecules by the combination of different atoms and, ••• to understand the origin and nature of the characteristics of electromagnetic radiation absorbed or emitted by atoms. 2.1 SUB-ATOMIC PARTICLES Dalton ' s atomic theory was able to explain

STRUCTURE OF ATOM

For your better entry test preparation and better Chemistry MCQs preparation, in this section, we are going to post Atomic Structure MCQs. An Atom is a small part of an element that takes part in chemical reactions. It is made up of three subatomic structures called Protons, Neutrons, and Electrons. Electrons are the smallest of the three particles that make up atoms.

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The central nucleus of the atom contains the neutrons and positively charged protons. The nucleus is surrounded by as many orbital electrons (each with a charge of -1) as there are. protons (each with a charge of +1) in the nucleus. The three sub-atomic particles differ from each other in two important ways.

Chapter 1 page 1 - 16 - University of Manchester

(a) atomic mass. (b) number of neutrons. (c) mass number. (d) atomic number Answer: (d) Atoms of the same element can differ in (a) chemical properties. (b) mass number. (c) atomic number. (d) number of protons and electrons. Answer: (b) The average atomic mass of an element is the average of the atomic masses of its (a) naturally occurring ...

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The element Z =114 has been discovered recently. It will belong to which of the following family/group and electronic configuration? (a) Halogen family, [Rn]5f 14 6d 10 7s 2 7p 5 (b) Carbon family, [Rn]5f 14 6d 10 7s 2 7p 2 (c) Oxygen family, [Rn]5f 14 6d 10 7s 2 7p 4 (d) Nitrogen family, [Rn]5f 14 6d 10 7s 2 7p 6

NEET and AIPMT NEET Chemistry Structure of Atom MCQ ...

MCQs on Structure of Atom for NEET. Everything around us is made up of atoms. Atom is the smallest unit of matter and has all the chemical properties of a matter. Atom has a positively charged nucleus, which contains positively charged particles known as protons and neutral particles called neutrons. The nucleus is surrounded by negatively charged electrons.