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a process by which particles  
of a gas spread out sponta-  
neously and mix with other  
gases. In contrast, effusion  
is a process by which gas  
particles pass through a  
tiny opening. The rates of  
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molecular theory. The theory

developed in the late 19th

century based on the idea of

the particles of matter are

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**Answers** in motion. ideal gas.  
and imaginary guess that  
perfectly fits all the  
assumptions of the  
Connecticut molecular  
theory.

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Characteristics Of Gases  
Answer Key For example, 10  
1, 10 2, 10 3, and so forth  
are all different orders of  
magnitude. All quantities  
that can be expressed as a  
product of a specific power  
of 10 are said to be of the  
same order of magnitude.

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*Answer Key*

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example,  $10^1$ ,  $10^2$ ,  $10^3$ ,  
and so forth are all  
different orders of  
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biological determinism  
principle that behavioral  
differences are the result  
of inherited physical  
characteristics gender  
identity a sense of being  
male or female based on  
learned cultural values and  
women in all societies would  
behave uniformly in their  
unique ways be-

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physical characteristics of  
the user, such as  
fingerprints or retinas  
Posted security guard  
Sensors, such as RFID tags,  
to monitor equipment ...

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the characteristics that  
enable the body to perform  
physical activity; more  
broadly, the ability to meet  
routine physical demands  
with enough reserve energy  
to rise to a physical  
challenge; or the body's  
ability to withstand stress

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**Gas** differences. ethnicity.  
social and cultural  
characteristics that set  
apart one group of people  
from another ... the process  
in which minority groups  
lose their distinct cultural  
...

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8 5 \_\_\_\_ 10 DIRECTIONS: Write

on the line at the right of

each statement the letter

preceding the word or

expression that best

completes the statement. 1.

According to the kinetic

theory, particles of matter

are in motion in (a) gases

only; (b) gases

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Review What are the major

characteristics of a plant



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