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~~STATES OF MATTER~~

~~LECTURE 1 AND 2~~

Chapter 10 - Gases

States of Matter

: Solid Liquid Gas

Change of State |

Matter | Physics |

FuseSchool *States*

of Matter - Solids,

Liquids, Gases

\u0026 Plasma -

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~~11th | States of~~

~~Matter | NCERT~~

~~Solutions: Q 1 to 15~~

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for Kids (Solid,

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of matter | States

of matter and

intermolecular

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Changes of States

of Matter GCSE

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~~of Matter \u0026amp;~~

~~Changing State~~

~~#20~~

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Differences

Between Solids,

Liquids and Gases

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States of Matter

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states of matter |

Bose-Einstein

condensates |

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~~15 States of Matter~~

~~Explained States of~~

~~Matter | #aumsum~~

~~#kids #science~~

~~#education~~

~~#children~~

~~Predicting States~~

~~Of matter in a~~

~~Chemical Reaction~~

~~| Identifying~~

~~liquid, solid, gas or~~

~~aqueous | VID19~~

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the Spadfoot Toad

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Solids, Liquids

\u0026 Gases |

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Matter | Chemistry

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Redefined Chem

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matter solid | liquid

| gases class 11

science

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of Matter. STUDY.

PLAY. Kinetic

Molecular Theory.

the theory that all

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matter is composed of particles moving constantly in random directions.

This theory explains the properties of what 3 things? Solids, liquids, and gases. Ideal gas.

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Section 1 The
Kinetic-Molecular
Theory of Matter

Section 2 Liquids

Section 3 Solids

Section 4 Changes
of State Section 5

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of Matter Section
10.1 The Nature of
Gases OBJECTIVES:
Describe the

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motion of gas particles according to the kinetic theory. Section

10.1 The Nature ...

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REVIEW States of
Matter SECTION 4

SHORT ANSWER

Answer the
following questions
in the space
provided. 1. a

When a substance
in a closed system
undergoes a phase

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State of Matter

Test
system reaches equilibrium, (a) the two opposing changes occur at equal rates. (b) there are no more phase changes. (c) one phase change predominates.

10 States of Matter

- Ms. Agostine's

Chemistry Page

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Modern Chemistry.

Chapter 10. States of Matter. Kinetic-Molecular Theory

The kinetic-molecular theory of matter is based on the idea that particles of matter are always in motion and this motion has consequences that affect its physical

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properties. Kinetic-Molecular Theory of Gases 1- Gases consist of large numbers of tiny particles that are far apart relative to their size. 2- Collisions between gas particles and between particles and their container walls are elastic.

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Modern Chemistry

Chapter 10 States of Matter

Almost all substances can be classified into three states of matter – solids, liquids and gases. Each state has different properties. Heating and cooling a substance can cause it to change

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test questions -

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Molecular Theory.

the theory that all matter is

composed of

particles moving

constantly in

random directions.

This theory

explains the

properties of what

3 things? Solids,

liquids, and gases.

Ideal gas.

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the kinetic theory

... The average

kinetic energy of all

the molecules in

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liquid water at...

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time we talk

concerning States
of Matter

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classifying matter

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of matter

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Answers

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of matter 1.

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of Matter • Kinetic
Molecular Theory:
the idea that
particles of matter
are always in
motion; this applies

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to all states of matter
matter • Used to explain properties of matter in terms of energy 2.

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SHORT ANSWER
Answer the

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following questions
in the space
provided Modern
chemistry chapter
10 review answers
states of matter
section 1. 1. Match
description on the
right to the correct
crystal type on the
left Modern
chemistry chapter
10 review answers
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Section 1. b ionic
crystal (a) has
mobile electrons in
the crystal c
covalent ...

Modern Chemistry
Chapter 10 Review
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Kinetic Energy Gas

Pressure Vacuum

Atmospheric

Pressure Va... 0

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CHEMISTRY NOTES

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- Chapter 10 States of Matter Goals : To gain an

understanding of :

- 1.. The kinetic theory of matter.
2. Avogadro's hypothesis.
3. The nature of liquids, solids and gases.
4. Phase changes.

NOTES: The kinetic theory of matter states that

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particles which
make up all types
of matter are in
constant motion.

CHEMISTRY NOTES

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of Matter

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