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Chemistry 146, Van Bramer, Chapter 15 lecture. Chapter 15 Lecture Outline. 15.3 Definitions of Acids and Bases The Arrhenius Definition Bronsted-Lowry Acid-Base Reactions. General Equation; Acid: Base: Conjugate Base of HA: Conjugate acid of B: HA + B ...

Van Bramer, Chemistry 146; Chapter 15 lecture outline

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CHEM201A Chapter 6 Lecture Outline Gottlieb 15 Example: Determine the mass of hydrogen gas, in g, that must be combusted to produce 1510.2 kJ of heat. $2\text{H}_2(\text{g}) + \text{O}_2(\text{g}) \rightarrow 2\text{H}_2\text{O}(\text{g})$ $\Delta H_{\text{rxn}} = -483.6 \text{ kJ}$ Example: A 32.3-mL sample of 0.996M NaOH is mixed with 19.1 mL of 0.905M H_2SO_4 .

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