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## Chapter 20 Acids Bases Practice Problems Answers

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10th Board Science !! 5 marks Important Question !! Ch -2 Acids, Bases & salts !! Vishal Tyagi ~~Table -~~

Equilibrium Constant Expression, Initial Concentration,  $K_p$ ,  $K_c$ , Chemistry Examples *Electrochemistry Review - Cell Potential & Notation, Redox Half Reactions, Nernst Equation* Modern Periodic Table **Acids and Bases, pH and pOH**

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Acid-Base Equilibrium Practice Problems Here are two hints about these problems in general: When it gives you a concentration of acid in the aqueous solution, that is your initial value for the equilibrium table (unless the problem specifically says that it is the equilibrium concentration). Also you can always calculate the equilibrium  $[H_3O^+]$  or  $[OH^-]$  from the pH.

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[Book] Chapter 20 Acids Bases Practice Problems Answers Practice-Problem-6660020 book December 22, 2010 8:37 Chapter 20: The Properties of Acids and Bases 61 where  $K_a = 7.0 \times 10^{-2}\text{ M}$  from Table 20.8. Solving for  $x$  using the method of successive approximations (about 7 or six iterations) or the quadratic formula gives

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Acid and Base Worksheet - Answers. 1) Using your knowledge of the Brønsted-Lowry theory of acids and bases, write equations for the following acid-base reactions and indicate each conjugate acid-base pair: a)  $\text{HNO}_3 + \text{OH}^- \rightarrow \text{H}_2\text{O} + \text{NO}_3^-$ .  $\text{HNO}_3$  and  $\text{NO}_3^-$  make one pair  $\text{OH}^-$  and

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H<sub>2</sub>O make the other. b) CH<sub>3</sub>NH<sub>2</sub> + H<sub>2</sub>O ( CH<sub>3</sub>NH<sub>3</sub><sup>+</sup> + OH<sup>-</sup>)

## Acid and Base Worksheet - Answers - Chemistry Made Easy

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CHAPTER 20. The Properties of Acids and Bases 20-1. Nitric acid is a strong acid, so  $[H_3O^+] = [NO_3^-] = 0.60 M$  and  $[OH^-] = K_w [H_3O^+] = 1.00 \times 10^{-14} M / 0.60 M = 1.7 \times 10^{-14} M$  20-2. Barium hydroxide, Ba(OH)<sub>2</sub>(aq), is a strong base, so  $[Ba^{2+}] = 0.25 M$   $[OH^-] = (2)(0.25 M) = 0.50 M$  and  $[H_3O^+] = K_w [OH^-] =$

## CHAPTER 20. The Properties of Acids and Bases

Acid Base Practice Test \_\_\_\_\_ 1. Acids taste a. sweet. c. bitter. b. sour. d. salty. \_\_\_\_\_ 2. Acids make litmus paper turn a. red. c. blue. b. yellow. d. black. \_\_\_\_\_ 3. Bases feel a. rough. c. slippery. b. moist. d. dry. \_\_\_\_\_ 4. Bases react with a. acids to produce salts and water. c. water to produce acids and salts. b. salts to produce acids and water. d. neither acids, salts, nor water. \_\_\_\_\_ 5.

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View Chapter 3 Practice Problems.pdf from CHEM 255 at University of Massachusetts, Boston. KEY Chapter 3 ACIDS and BASES 1. For the following reaction, which reactant functions as an acid? 2. For the

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