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Chapter 24 (Organic Chemistry) -

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Part 1 Chapter 22 (Organic  
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Chemistry CHAPTER 22 REVIEW  
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SHORT ANSWER Answer the  
following questions in the space

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provided. 1. Match the reaction type on the left to its description on the right. \_\_\_\_\_ substitution  
\_\_\_\_\_ addition \_\_\_\_\_ condensation  
\_\_\_\_\_ elimination 2. Substitution reactions can require a catalyst to be feasible. The reaction represented by the ...

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Last updated 3/28/2019 1 The  
previous three chapters were  
about adding things directly to

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Answers Modern the base of a carbonyl. In this chapter, we'll look at reaction that happen on the carbon next to the base of the carbonyl, the  $\alpha$  carbon.

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3/28/2019 4 Usually the  
expectation is that if heat is

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written on the arrow or if the product has some big benefit from being conjugated (for example see Claisen-Schmidt condensation below), you'll get the condensation product.

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Halogenation This just means  
removing an  $\alpha$  H and attaching a

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halogen in its place. You can do this in either acid or base. In acid, you make the enol first, by the mechanism on the first page.

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X2, hv Predicting product: Identify which carbon could give the most stable radical, and substitute a Br for an H on that carbon.

Stereochemistry: Leads to racemic, due to achiral radical ...

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and how reactions occur is critically important to understanding the processes in an industrial plant. For biologists and health professionals, it is essential because nearly all of biochemistry springs from organic chemistry. Additionally, all

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"protecting groups" which they attach to functional groups (reactive groups of atoms) while some reaction is being conducted on another part of the molecule. These protecting groups prevent reactions of the functional groups during other reactions and are

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removed later by a

heterogeneous catalytic method called hydrogenolysis. One unique feature of this book, not found in other books on catalysis, is an exhaustive chapter (Chapter 4) on hydrogenolysis, which is dredged from the recent synthetic

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selective low temperature, liquid phase heterogeneous catalytic oxidations by hydrogen peroxide (Chapter 6). These specialty topics are timely and may be new to academic organic chemists and can be used to supplement their advanced courses. Several

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informed by their supervisor that they must convert some of their synthetic steps into heterogeneous catalytic steps. They may not have been exposed to heterogeneous catalysis and have few places to turn. This book offers them a crash course in

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