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compound. a
substance composed
of more than one type
of atom linked
together chemically.

colloid. a mixture
containing tiny clumps
or particles that
remain suspended

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within the mixture .
solution. a uniform
mixture in which one
substance is
dissolved in another
substance. empirical
formula. a type of
formula that ...

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Section 6.4 The
Structure of Metals
(pages 176–181) This
section discusses
metallic bonds and
the properties of
metals. It also
explains how the
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potential energy.
kinetic energy.
mechanical energy.
the ability to do work.
stored energy that

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results from the position or shape of an object. the energy an object has due to its motion. the energy associated with the motion and position of everyday objects.

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atom least likely to
react? When the
highest energy level is
filled with electrons,
an atom is stable and
not likely to react.
Describe one way an
element can achieve
a stable electron

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Study the electron dot
diagrams for lithium,
carbon, flourine, and
neon in figure 6-1.

Choose the statement
that correctly
identifies the most

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stable of the
elements. A) lithium is
the most stable
element because it
has to lose only one
electron to achieve a
stable configuration.
B) carbon is the most
stable element
because it can form
four bonds.

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science AR □□When

two atoms of the
same nonmetal react,
they often form a(an)
Diatomic molecule

The formation of an
ionic bond involves
the transfer of
electrons

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