

## Chapter 7 Ionic Metallic Bonding Work Answers

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Section 7.3 – Bonding in Metals The valence electrons of metal atoms can be modeled as a sea of electrons. Metallic bonds consist of the attraction of the free-floating valence electrons for the positively charged metal ions. Metals are good conductors and malleable because of their mobile electrons.

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Ionic Bonding. Anions and cations are held together by . opposite. charges (+ and -) Ionic compounds are called . salts. Formula unit – Simplest . ratio. of elements in an ionic compound. Practice with Formula Unit. ... Chapter 7 Ionic and Metallic Bonding Last modified by:

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Chapter 7: Ionic and Metallic Bonding. valence electrons. electron dot structures. octet rule. halide ions. electrons in the highest occupied energy level of an element's.... diagrams that show valence electrons in the atoms of an elemen.... in forming compounds, atoms tend to achieve the electron confi....

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OBJECTIVES:-Determine the number of valence electrons in an atom of a representative element. 3

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Chapter 7 - Ionic/Metallic Bonding Vocabulary 1. Anion- an ion with a negative electrical charge 2.

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Cation-ion with a positive electrical charge. 3. Chemical Bond- is a lasting attraction between atoms, ions or molecules that enables the formation of chemical compounds. 4.

*Jude Pena - Chapter 7 - Ionic\_Metallic Bonding Vocabulary ...*

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Metallic Bonds and Metallic Properties 1. Is the following sentence true or false? Metals are made up of cations and valence electrons, not neutral atoms. 2. What are metallic bonds? 3. Name three properties of metals that can be explained by metallic bonding. a. b. c. 4. What happens to an ionic crystal when a force is applied to it? 5.

***BONDING AND INTERACTIONS***

Ionic Bonding; Chapter 6 - Chemical Bonding; Ionic Charges; Structure & Properties of Ionic & Covalent Compounds; Covalent, Ionic, and Metallic bonds; Ionic Compounds 2; IONIC

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