

Chapter Test B Gases

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_____ 1. According to the kinetic-molecular

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theory, particles of an ideal gas a. attract each other but do not collide. b.

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the sum of the partial pressures of the component gases. 2. Chapter Test B Gases Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1.

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or phrase that best completes each statement or best answers each question. _____ 1. If the temperature of a gas remains constant, then the pressure of the gas will increase if the Assessment Chapter Test B

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Chemistry Chapter 14 The Behavior of Gases Test. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Aubr3y_Davis. Terms in this set (10) at constant volume and temperature the total pressure exerted by a mixture of gases is equal to the sum of partial pressures of the component gases.

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(b) An English unit of pressure sometimes used in engineering is pounds per square inch (lb/in. ²), or psi: 1 atm = 14.7 lb/in. . If a pressure is reported as 91.5 psi, express the measurement in atmospheres. Answer: (a) 99.3 kPa, (b) 6.22 atm Practice Exercise 1) A gas at a pressure of 10.0 Pa exerts a force of _____ N on an area of A) 55

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