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1. Study the flow chart below and answer the questions that follow: 4. Name reagent Z. 5. Describe the process which takes place in step 2. 6. Identify the white solid. 2. a) Starting from solid magnesium oxide, describe how a solid sample of magnesium hydroxide can be prepared. b) Give one use of magnesium hydroxide. 3.

~~QUESTION — CHEMISTRY FORM 2~~

FORM 2 CHEMISTRY TIME: 2 Hrs INSTRUCTIONS Attempt all the question in the question in the spaces provided All the working must be clearly shown where appropriate. Negligence will be penalized. QUESTIONS 1. The table below shows some information concerning partials of certain elements. Element Atomic number Number of electrons Mass Number P 19 18 39

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1.1 The structure of the atom. When scientists started exploring matter, they realised that matter can be divided into smaller and still smaller particles. They called the smallest particle an 'atom'. The name 'atom' was derived from the Greek word 'atomos', meaning ...

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FORM 2 CHEMISTRY 2013 INSTRUCTIONS TO CANDIDATES

1. Answer all the questions in the spaces provided. 2. Mathematical tables and electronic calculators may be used. 3. All working must be clearly shown where necessary 1. The table below shows the pH values of solutions J to N.

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The balanced chemical equation is: $\text{CaCO}_3(\text{s}) + 2 \text{HCl}(\text{aq})$ to

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$\text{H}_2\text{O}(\text{l}) + \text{CaCl}_2(\text{aq}) + \text{CO}_2\text{...}$ View Answer Potassium chlorate, KClO_3 , is physically characterized as a white crystalline or powdery...

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