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answers in the boxes provided. 1. Ethanedioic acid is a diprotic acid. A student determined the value of  $x$  in the formula of hydrated ethanedioic acid,  $\text{HOOC}-\text{COOH}\cdot x\text{H}_2\text{O}$ , by titrating a known mass of the acid with a  $0.100 \text{ mol dm}^{-3}$  solution of  $\text{NaOH}(\text{aq})$ .  $0.795 \text{ g}$  of ethanedioic acid was dissolved in distilled water and made up to a total ...

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5. a ratio of electrons : aluminium ions  $3 : 1$  amount Al  
 $13 \frac{1}{2} \times 27 = 35.25 \text{ g}$   
mass Al  $\llcorner 35.25 \text{ g} \llcorner 26.98 \text{ g mol}^{-1} \llcorner 1.296 \text{ mol}$   
Award [3] for correct final answer. 3 5.

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Common errors students make when studying are:  
#1: Avoiding the topics you didn't fully comprehend in  
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The Best IB Chemistry Study Guide and Notes for  
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