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Chapter 4 Reactions in Aqueous Solution (Sections 4.1 - 4.4)

Chapter 2 The Chemical Level of Organization Acid-Base

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2. Atomic Structure

Which way will the Equilibrium Shift? (Le Chatelier's Principle)Le Chatelier's Principle

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Total mass before reaction: $18.00 \text{ g} + (25.00 \times 1.025 \text{ g/mL}) = 43.63 \text{ g}$

g After reaction following the law of conservation of mass: 43.63 g

$= 12 \text{ g} + 30.95 \text{ g} + \text{mass of H}_2$ mass of $\text{H}_2 = 0.68 \text{ g}$; volume of H_2

$= 0.68 \text{ g} \times 1 \text{ L} / 0.0824 \text{ g} = 8.25 \text{ L}$ Solutions Manual for Chemistry

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