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53. ChemQuest 17.

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Information: Separating
Charges. Examine
Figure 1 below where
there are three pairs of
metal spheres that have
different amounts of
charge on them. The
spheres in diagram C

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are closer than the others because they have the strongest attraction.

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to help you complete the

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molarity chemquest
found in Unit 11.

Molarity

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Name: _____ Date:

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Information: Molarity
Concentration is a term
that describes the
amount of solute that is
dissolved in a solution.
Concentrated solutions
contain a lot of

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Answers dissolved solute, but dilute solutions contain only a little.

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An element exists as 4
different isotopes.
4.35% have a mass of
49.9461, 83.79% have a
mass of 51.9405, 9.50%
have a mass of 52.9407

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amu, and 2.36% have a mass of 53.9389 amu. a. What is the average atomic mass for this element? b. What is the identity of this element?

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Key More Hint: add

your answer to question

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Answers
1 and four times your
answer to question 2. 4
 $+ 4(7) = 4 + 28 = 32$ 4.

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Key More Hint: add

your answer to question

1 and four times your

answer to question 2. 4

+ 4(7) = 4 + 28 = 32 4.

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you in imitation of ...

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15. Given your answer to question 14 and the reaction you wrote in question 13, calculate the molarity of the nitrate ions (NO_3^-) in the solution. Title:

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(Base the answer to this question on the number of moles of propane that actually get

combusted—which is your answer to part a.)

12 moles. For every mole of propane that combusts 3 moles of

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CO₂ are produced, so the number of moles of CO₂ that can be produced when 4 moles of propane combusts = $4(3) = 12$.

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