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Cathodic Protection Four parts needed for a CP cell to exist 1. Anode –Where corrosion occurs 2. Cathode –Protected from corrosion 3. Electrolyte Soil or water (any conductive environment) adjacent to –and containing both the anode and the cathode 4. Metallic Path Physically connects the

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Corrosion.ppt 1. PREPARED BY: DARSHAN R. KANJIYA 11MCY 043 KSKV KACHCHH UNIVERSITY, DEPARTMENT OF CHEMISTRY, BHUJ 1
2. 2 3. INTRODUCTION Corrosion is a process of formation of the compound of pure metal by the chemical reaction between metallic surface and its environment. It is an oxidation process.

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rotective Measures Against Corrosion Organic Coating : Apply on metallic surface for protection from corrosion & also to impart decorative value. Such as paints, enamel, Varnishes & lacquers Corrosion Inhibitors Substances which when added in a small amount in an environment reduces the rate of corrosion of a metal exposed to that environment.

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corrosion protection for a wide variety of environmental conditions. Nevertheless, the choice of the material or the corrosion protection ... We only talk about corrosion when there is a change in the metal ' s or system ' s properties which may lead to an undesirable outcome. This can range simply from

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Our Guides to Good Practice in Corrosion Control provide advice concerning corrosion control, including corrosion protection schemes and corrosion management for a variety of specific applications in the construction industry.

Corrosion good practice guides - NPL

This chapter presents important aspects of corrosion in industrial infrastructure, its causes, impacts, control, protection and prevention methods. 2. Materials in industry Metallic materials play a key role in the development of a country and its sustained growth in the context of the global economy.

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