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equilibrium if the indicated stress is applied to the reaction system. Mrs. Zuberbuehler - Mrs. Zuberbuehler

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Dynamic Equilibrium At what point is a reversible reaction "completed?" Most of the reactions we have studied year have been "forward" reactions—once the reactant has been changed into the product it stays that way. The reaction "stops" when the limiting reactant runs out.

~~4—SCIENCE~~

Increase (T) causes the equilibrium to shift to the right (+)

Decrease (I) causes the equilibrium to shift to the left

Increase (T) causes the equilibrium to shift to the left

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Decrease (I) causes the equilibrium to shift to the right
Temperature: A change in temperature corresponds to a change in energy therefore by using

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