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This excellent new text dispels the fear that the word electrochemistry commonly instils in chemistry students. Throughout the mathematical content has been left to a minimum for clarity, whilst retaining the important necessary physical insight.

This textbook provides a framework of the key concepts involved in electrochemical kinetics. A wide range of modern electrochemical techniques and applications are discussed. The mathematical content has been minimized for clarity, while retaining the important results necessary for physical insight. A substantial series of examples and illustrations is taken from the recent research literature to explore the potential applications of electrochemical techniques.

Another winning primer! This new addition to the popular series provides a basic introduction to equilibrium electrochemistry, focusing on electrode potentials and their applications. It builds on a knowledge of elementary thermodynamics giving the student an appreciation of the origin of electrode potentials and shows how these are used to deduce a wealth of chemically important information and data such as equilibrium constants, the free energy, enthalpy and entropy changes of chemical reactions, activity coefficients, the selective sensing of ions. It is mathematically simple, the emphasis throughout is on understanding the foundations of the subject and how it may be used to study problems of chemical interest.

The renowned Oxford Chemistry Primers series, which provides focused introductions to a range of

important topics in chemistry, has been refreshed and updated to suit the needs of today's students, lecturers, and postgraduate researchers. The rigorous, yet accessible, treatment of each subject area is ideal for those wanting a primer in a given topic to prepare them for more advanced study or research. Moreover, cutting-edge examples and applications throughout the texts show the relevance of the chemistry being described to current research and industry. The learning features provided, including questions at the end of every chapter and online multiple-choice questions, encourage active learning and promote understanding. Furthermore, frequent diagrams, margin notes, further reading, and glossary definitions all help to enhance a student's understanding of these essential areas of chemistry. This brand new addition to the series provides the most accessible first introduction to electrochemistry, combining explanation of the fundamental concepts with practical examples of how they are applied in a range of real-world situations.

Comprehensive text and reference covers all phenomena involving light in semiconductors, emphasizing modern applications in semiconductor lasers, electroluminescence, photodetectors, photoconductors, photoemitters, polarization effects, absorption spectroscopy, more. Numerous problems. 339 illustrations.

This book will give students a thorough grounding in pH and associated equilibria, material absolutely fundamental to the understanding of many aspects of chemistry. It is, in addition, a fresh and modern approach to a topic all too often taught in an out-moded way. This book uses new theoretical developments which have led to more generalized approaches to equilibrium problems; these approaches are often simpler than the approximations which they replace. Acid-base problems are readily addressed in terms of the proton condition, a convenient amalgam of the mass and charge constraints of the chemical system considered. The graphical approach of Bjerrum, Hagg, and Sillen is used to illustrate the orders of magnitude of the concentrations of the various species involved in chemical equilibria. Based on these concentrations, the proton condition can usually be simplified, often leading directly to the value of the pH. In the description of acid-base titrations a general master equation is developed. It provides a continuous and complete description of the entire titration curve, which can then be used for computer-based comparison with experimental data. Graphical estimates of the steepness of titration curves are also developed, from which the practicality of a given titration can be anticipated. Activity effects are described in detail, including their effect on titration curves. The discussion emphasizes the distinction between equilibrium constants and electrometric pH measurements, which are subject to activity corrections, and balance equations and spectroscopic pH measurements, which are not. Finally, an entire chapter is devoted to what the pH meter measures, and to the experimental and theoretical

uncertainties involved.

Electron transfer reactions are of fundamental significance in many areas of inorganic, organic and biological chemistry, and electrochemical techniques are a useful tool for studying them. This book provides an overview of recent advances in voltammetry and electrochemistry, broadening the scope of their application and suggesting new problems that they may be able to address in the 21st century.

Showing how to apply the theoretical knowledge in practice, the one and only compilation of electrochemical experiments on the market now in a new edition. Maintaining its didactic approach, this successful textbook provides clear and easy-to-follow instructions for carrying out the experiments, illustrating the most important principles and applications in modern electrochemistry, while pointing out the potential dangers and risks involved. This second edition contains 84 experiments, many of which cover electrochemical energy conversion and storage as well as electrochemical equilibrium.

This is an introduction to the areas of application of electroanalysis, which has an important role with current environmental concerns, both in the laboratory and in the field.

New edition of the overwhelmingly favorite text for the physical chemistry course.

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