

## Electrolyte Solutions Robinson Stokes

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*Electrolyte Solutions*, von R. A. Robinson und R. H. Stokes. - The Measurement and Interpretation of Conductance, Chemical Potential and Diffusion in Solutions of Simple Electrolytes. Butterworths Scientific Publication, London 1959. 2.

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Stokes and Robinson (1) in 1948 presented a simple theory of electrolyte hydration and derived an equation by which to calculate activity coefficients for electrolytes. The equation contains two adjustable parameters, the effective size and the hydration number of the electrolyte.

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Semi-empirical corrections for concentrated electrolyte solutions For  $I < 0.1 \text{ mol kg}^{-1}$ , For  $I < 0.1 \text{ mol kg}^{-1}$ ,  $b$  is an adjustable parameter . 9.8 Weak Electrolytes Weak Electrolytes are compounds (such as acetic acid) that are only partially dissociated in aqueous solutions. At equilibrium, in addition to the ions, there exists a significant concentration of the molecular (undissociated ...

*CH. 9 Electrolyte Solutions*

Buy Electrolyte Solutions: Second Revised Edition (Dover Books on Chemistry) on Amazon.com FREE SHIPPING on qualified orders ... What is most useful is the large appendix at the end which is effectively a collection of data collected by Robinson and Stokes throughout their career. All of that data can be obtained for free at any university by finding their original publications, but for the ...

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rR0blflS0flR A &StokesR H. Electrolyte solutions. The measurement and interpretation of conductance, chemical potential and diffusion in solutions of simple electrolytes. London: Butterworths, 1959. 571 p.

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