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heat capacities of various elements and compounds are C_p Ags = $C_p \hat{e}$. 8a Ø 21.30, b Ø 8.54, c Ø 1.51 ...

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DH . Work is found thethe first law as $w = q - \Delta U$; thus $q = \Delta H$; $w = \Delta H - \Delta U$; 4.

Isothermal Process Because U is a function only of T for an ideal gas, $\Delta U = \Delta H = 0$ for an isothermal process. These

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plot of $V = V(P, T)$ for a gas is shown in

Fig 11

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