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The formula is: $M_1 V_1 = M_2 V_2$ Where M_1 and V_1 are the molarity and volume before a dilution, and M_2 and V_2 are the molarity and volume after a dilution. Consider this example from the 2006 NECTA: "Hydrochloric acid solution made by diluting 750 cm³ of 0.25 M HCl to 937.5 cm³ with distilled water."

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Guide To Chemistry Practicals Maktaba

This material has been written to support students with the practical work of Advanced Higher Chemistry and in particular the Researching Chemistry unit, which includes the investigation. It is divided into four main sections:

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