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understanding
the chemical
process then
designing,
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The kinetic and the potential energy are

negligible in many chemical reaction

engineering

applications, so

Equation 15

becomes $Q - V = E = Q$

$V (U + K + E + P) \approx$

$Q - V = U$ (16) we

know that U is a

function of the

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enthalpy,
pressure, and
volume, so $Q = V$

$$U = Q - V \left(H - PV \right) =$$
$$Q - V \sum_{i=1}^n X_i C_i \left(h_i^h - P_i^h \right)$$
$$= \sum_{i=1}^n X_i C_i \left(F_i^h - P_i^h \right) \quad (17)$$

When this is applied for both streams, the term P_i^h cancels with the volumetric work

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is a change in
the atomic
content, or
atomic

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configuration,
of a molecule of
a particular
chemical species
brought about by
interaction with
molecules of
other chemical
species, or due
to an input of
energy, a
chemical
reaction takes
place! Such

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transformations
form the basis
for manufacture
of new
chemicals,
pharmaceuticals,
fuels and
materials and
also occur in
nature.

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by Lord Kelvin
by combining
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and power
(dynamis) The
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“separation
processes”?

Separations are
crucial in
chemical engi-
neering (e.g.,
chemical plants,
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refineries)

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commonly have
from 40% to 70%
of both capital

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