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Numerical Answers. According to Equation 9.1, in the first case $Q_1Q_2 = (+1) (-1) = -1$; in the second case, $Q_1Q_2 = (+3) (-1) = -3$. Thus, E will be three times larger for the $+3/-1$ ions.

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Size) In ionic bonding, the
atom of the element is
bonded to an atom of the
opposite sign. For example,

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boron atoms bonded to oxygen
atoms are covalent.

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Ionic And Covalent Bonds

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each pair of elements below
draw an atomic diagram
showing electrons in

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different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction.

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Answers
source:mychaume.com A
solvent is a thing which
could dissolve different
things. It may, therefore,
be presumed this might be a
reason as to why ionic and
covalent substances aren't
excellent conductors of

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