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Terms in this

set (32) Which

one of the

following

reactions is NOT

balanced? $2 \text{CO} +$

$\text{O}_2 \rightarrow 2 \text{CO}_2$ 2SO_2

$+ \text{O}_2 \rightarrow 2 \text{SO}_3$ SF_4

$+ 3 \text{H}_2\text{O} \rightarrow \text{H}_2\text{SO}_3$

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ANSWER: Correct

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Reading Question

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reacts with

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oxygen in the

air to produce

rust (Fe_2O_3)

according to the

reaction shown

below. $4 \text{Fe} + 3$

$\text{O}_2 \rightarrow 2 \text{Fe}_2\text{O}_3$ How

many moles of

rust can be

formed if 1.8

mols of iron are

present? Hint 1.

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ANSWER Answer

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the following

questions in the
space provided.

1. List four
metals that will
not replace
hydrogen in an
acid. Choose
from Cu, Ag, Au,
Pt, Sb, Bi, and
Hg.

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Chapter 2: Rank

from most to

fewest valence

electrons of the

following

elements: Kr,

Ga, Br, As, K. -

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Answer: Kr (8),
Br (7), As (5),
Ga (3), K (1).

Valence

electrons are
equal to the
atom's main
group number.

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table, one of
the columns
within the main
group elements;

a family or

group of

elements

exhibits similar

chemical

properties.

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