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5: The Periodic Law. STUDY. PLAY.

Periodic law. The physical and chemical properties of the elements are periodic functions of their atomic numbers. Periodic table. An arrangement of the elements in order of their atomic numbers so that elements with similar properties fall in the same column, or group.

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Highlights 5.1 - The periodic law states that the physical and chemical properties of the elements are periodic functions of their atomic numbers. - The periodic table is an arrangements of the elements in order of their atomic numbers so that elements with similar properties fall in

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Chapter 5: The Periodic Law Terms in
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section 1 (history of the periodic
table). The students complete 25 fill in
the blank questions with words from
the chapter as they read. There are
also 2 extra credit. A key is included.

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13. c 14. d 15. c 16. b 17. d 18. a 19.
d 20. c 21. d 22. a 23. a 24. d 25. b

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developing the first successful a.
periodic table. b. method for
determining atomic number. c. test for
radioactivity. d. use of X rays. ____ 6.
Mendelev did not always list
elements in his periodic table in order
of increasing atomic mass because he
grouped together elements with
similar. a. properties. c. densities. b.

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of the ability of an atom in a chemical
compound to attract electrons from
another atom in the compound is
called _____. 17. The energy required
to remove one electron from an atom
is called its _____. 18.

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