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formula.

indicates the

relative number

of atoms of each

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kind in a  
chemical  
compound.

Reveals the

number of atoms

of each element

contained in a

single molecule

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SECTION 1 SHORT  
ANSWER Answer

the following  
questions in the  
space provided.

1. c In a Stock  
system name such  
as iron(III)  
sulfate, the



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Roman numeral

tells us (a) how many atoms of Fe are in one

formula unit.

(b) how many sulfate ions can be attached to the iron atom.

(c) the charge on each Fe ion.

*7 Chemical*

*Formulas and*

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REVIEW SHORT

ANSWER Answer

the following

questions in the

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space provided.

1. Write formulas for the following

compounds:  $\text{CuCO}_3$

a. copper(II) carbonate  $\text{Na}_2\text{SO}_3$

b. sodium sulfite  $(\text{NH}_4)_3\text{PO}_4$

c. ammonium

phosphate  $\text{SnS}_2$

d. tin(IV) sulfide  $\text{HNO}_2$

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nitrous acid 2.

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associated with

how electrons

are arranged in

atoms.

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SECTION 1 SHORT

ANSWER Answer

the following

questions in the

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1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number.

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(d) based on  
when they were  
discovered. 2. d  
Mendeleev  
noticed that  
certain  
similarities in  
the ...

Answers

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