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~~Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. ____ In metals, the valence electrons are considered to be (a) attached to particular positive ions.~~

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~~CHAPTER 9 REVIEW Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N₂ are mixed with 12.0 mol of H~~

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~~Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. ____ In metals, the valence electrons are considered to be (a) attached to particular positive ions.~~

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~~ANSWER KEY 1 Modern chemistry answer key chapter 2. a 2. b 3. c 4. c 5. a 6. b 7 Modern chemistry answer key chapter 2. c 8. Polarity in a water molecule is caused by an uneven distribution of electrons between the oxygen and hydrogen atoms. 9. The concentration of H⁺ ions determines whether a solution is acidic or basic.~~

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~~The answers to the Hold chemistry chapter 12 section 1 review can not be located online. Students will have to obtain assistance from the teacher if they need help with the answers.~~