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equation? 40.07 g/mol b. What is the molar mass of C_3H_4 ? 2 mol
O 2:1 mol H 2O c. What is the mole ratio ...

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Modern Chemistry 77 Stoichiometry CHAPTER 9 REVIEW

Stoichiometry SECTION 3 PROBLEMS Write the answer on the line to the left. Show all your work in the space provided. 1. _____
The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of N_2 are mixed with 12.0 mol of H

CHAPTER 9 REVIEW Stoichiometry

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CHAPTER 9 Stoichiometry Stoichiometry comes from the Greek

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words stoicheion, meaning "element," and metron, meaning "measure." Back Introduction to Stoichiometry SECTION 9-1 OBJECTIVES Define stoichiometry. The reaction-stoichiometry problems in this chapter can be...

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