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Membrane lipids.

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Covalent Bond.

Molecule. Diatomic
molecule. Molecular
compound. A bond
formed by the sharing of
electrons between

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atoms. A neutral group of atoms joined together by covalent bonds. A molecule consisting of two atoms.

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molecular structure of

water shows how the

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How are atoms joined

together to make

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Covalent bond.

Molecule. Diatomic

molecule. Molecular

compound. bond formed

by the sharing of

electrons between

atoms. a neutral group

of atoms joined together

by covalent bonds. a

molecule that contains

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two atoms. a compound
composed of molecules.

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share electrons. •There

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are several electrostatic

interactions in these

bonds: Attractions

between electrons and

nuclei, Repulsions

between electrons,

Repulsions between

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Reserved. In general,
covalent bonding results
from an imbalance

between the attractions
and repulsions of the
nuclei and electrons
involved. • The nuclei
and electrons attract
each other. • Nuclei
repel other nuclei.

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*Covalent Bonding -
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Chapter 8 Covalent
Bonding 183 Section
Review Objectives •

State a rule that usually
tells how many

electrons are shared to
form a covalent bond •

Describe how electron
dot formulas are used •

Predict when two atoms

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are likely to be joined
by a double or a triple
covalent bond •

Distinguish between a
single covalent bond
and other covalent
bonds

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8.4 Polar Bonds and

Molecules > - Pittsfield

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form of chemical

bonding between two

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non metallic atoms

which is characterized
by the sharing of pairs
of electrons between

Bonding

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Octet Rule4. • Atoms

tend to gain, lose, or

share electrons until

they are surrounded by

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eight valence electrons;
this is known as the
octet rule. • An octet
consists of full sand
psubshells. • We know
that ns^2np^6 is a noble
gas configuration.

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