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4.2.2 How much energy is released
when bonds form to produce 2 mol of
H₂O and 1 mol of CO₂? (4) 4.2.3

Hence calculate the energy transferred
(ΔH) for the combustion of 1 mol CH₄.

(2) 4.2.4 Draw a labelled energy profile
for this reaction. Include the labels:

ENTHALPY CHANGE (ΔH)

ACTIVATION ENERGY (E_a) (6)

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