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Nuclear Chemistry Review. Show all questions <= ... Identify the missing particle in the following nuclear reaction: ${}^{99}_{43}\text{Tc} + \text{_____} + 0^{-1}\text{e}$? ${}^{99}_{44}\text{Ru}$? ${}^{99}_{43}\text{Tc}$? ${}^{99}_{42}\text{Mo}$? ${}^{99}_{43}\text{Es}$; What is the effect of this decay? ${}^{99}_{43}\text{Tc} + 0^{-1}\text{e}$? a neutron is converted to a proton ?

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In this case, the particle's mass number is 0 and its atomic number is -1, indicating that the missing particle is a ${}^0_{-1}e$ particle, which is an electron emitted by an unstable nucleus. A beta particle is written as ${}^0_{-1}e$. Therefore, the final form of this nuclear particle is ${}^0_{-1}e$ and the full nuclear equation is

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The nuclear fuel cycle comprises the handling of all fissile and fertile material necessary for nuclear power production and of the radioactive products formed in this process (Fig. 21.1). The fuel cycle is suitably divided into a front end and a back end part, where the nuclear power station is the dividing line.

Radiochemistry and Nuclear Chemistry | ScienceDirect

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