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Solution:  $p_{abs} = p_{atm} + p_{gage} = 101.3 \text{ kPa} + 20 \text{ in. H}_2\text{O} = 110.26 \text{ kPa}$  Ideal gas law:  $\rho = \frac{p}{RT} = \frac{110.26 \text{ kPa}}{1 \text{ kg/K} \cdot 2077 \text{ J/K} \cdot 293.2 \text{ K}} = 0.181 \text{ kg/m}^3$  b.) Situation: A sphere of 93 mm diameter contains an ideal gas.  $T = 20 \text{ C} = 293.2 \text{ K}$  Find: Calculate the density of argon at a vacuum pressure of 8.8 psi. Properties:

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Solution 1.1. To get started, first list or determine the volumes involved:  $d = \text{volume of water dumped} = 100 \text{ cm}^3$ ,  $c = \text{volume of a sip} = 5 \text{ cm}^3$ , and  $V_2 = \text{volume of water in the oceans} = \frac{4}{3} \pi R^3 D$ , where,  $R$  is the radius of the earth,  $D$  is the mean depth of the oceans, and  $f$  is the oceans' coverage fraction.

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Solution We are to define the Mach number of a flow and the meaning for a Mach number of 2. Analysis The Mach number of a flow is defined as the ratio of the speed of flow to the speed of sound in the flowing fluid. A Mach number of 2 indicate a flow speed that is twice the speed of sound in that fluid.

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SOLUTION  $p=0$ . The fluid is considered compressible.  $dp_{EV} = \rho g dz$  However,  $V = dV > V$ .  $m$ . Then,  $r \cdot z = h z - (\rho g r^2) dp$ .  $dV = V \cdot dr = -\rho g r \cdot dr$ . Therefore,  $EV = \rho \cdot dp \cdot dr > r$  (a) At the surface, where  $p \dots$

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