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TEACHING TRANSPARENCY WORKSHEET Atomic Orbitals 1. What is the shape of an s orbital? Class 16 Use with Chapter 5, Section 5.2 2. What is the relationship between the size of an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a ublevel? 4.

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1. calcium and oxygen 2. Yes; each contains equal numbers of protons and electrons. 3. ionization energy 4. positive; cation 5. negative; anion 6. No; it is not a stable octet of electrons. 7.

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