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Magic - Boyle's Law |

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~~Combined \u0026amp; Ideal~~

~~Density, Molar Mass,~~

~~Mole Fraction, Partial~~

~~Pressure, Effusion Gas~~

~~Laws - Equations and~~

~~Formulas Gas Laws~~

How to Use the Ideal

Gas Law in Two Easy

Steps

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Easy way to Remember

Gas Law Equations

Kinetic Molecular

Theory and the Ideal

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Science at Home - SE3 -

EP6: Egg in a Bottle -

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Explaining the Gas

Laws in Chemistry -

Volume, Temperature,

Pressure, Moles....Made

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Gas Stoichiometry

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Chemistry #13 Dalton's

Law of Partial Pressure

Problems \u0026amp;

Examples - Chemistry

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~~Water Practice~~

~~Problems - Chemistry~~

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~~Chemistry Gas Laws~~

~~Magic~~

Technical Chemistry:

Gas Laws Name: Match

the variables used to



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describe gases to the correct unit. 1. 2. 4. 5  
kPa mL K mm Hg  
atmospheres (atm) L a.  
pressure b. temperature  
c. volume Complete the following statements by writing "decreases," "increases," or "remains the same" on the line provided. As a gas is compressed in a cylinder  
9. its mass

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~~Region 14 - Bethlehem  
& Woodbury  
Connecticut~~

Technical Chemistry -  
Gas Laws Magic Square

You must show your  
work in the square.

Name A. A sample of  
neon gas occupies a  
volume of 2.8 L at 1.8  
atm. What would its  
volume be at 1.2 atm?

B. A balloon full of air  
has a volume of 2.75 L

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at a temperature of  
18 ° C. What is the  
balloon's volume at 45  
° C? If 3.0 L of a gas at  
heated to 30.0 ° C

~~0 3L - Ms Galloway~~  
Bronwyn Hogan May  
10, 2001 Technical  
Chemistry - Gas Laws  
Magic Square You must  
show your work in the  
square. Name & mldr; &  
mldr; & mldr; & ml

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mldr; A. A sample of

neon gas occupies a

volume of 2.8 L at 1.8

atm. What would its

volume be at 1.2 atm?

B. A balloon full of air

has a volume of 2.75 L

at a temperature of 18 o

C.

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C. If 3.0 L of a gas at  
20.0 °C is heated to 30.0

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oc what is the new  
volume of the gas? (3 D

'2-1 9. 11.3L A. A

sample of helium gas  
occupies a volume of 4.5  
L at 5.8 atm.

~~Gas Laws Magic~~

~~Squares Answer Key~~

~~Weebly~~

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Gas Laws Magic Square

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work in the square. Na

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mldr;&mldr;&mldr;&ml  
dr;&mldr;&mldr; A. A  
sample of neon gas  
occupies a volume of 2.8  
L at 1.8 atm. What  
would its volume be at  
1.2 atm? B. A balloon  
full of air has a volume  
of 2.75 L at a  
temperature of 18 o C.

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Magic Squares 1 answer

below » 1. A Sample of

neon gas occupies a

volume of 2.8 L at 1.8

atm. What would its

volume be at 1.2 atm?

Feb 11 2011 07:15 PM.

1 Approved Answer.

Mark B answered on

February 12, 2011. 3

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full of air has a C. If 3.0

L of a gas at  $20.0^{\circ}\text{C}$  is

occupies a volume of 2.8

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A problem  
A volume of 2.75 L at a  
heated to  $30.0\text{ }^{\circ}\text{C}$  what  
is at 1.8 atm. What  
would its volume be at  
1.2 atm? temperature of  
 $18\text{ }^{\circ}\text{C}$ . What

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20.0 oc is heated to 30.0



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oc what is A. A sample of neon gas occupies a volume of 2.8 L at 1.8 atm. What would its volume be at 1.2 atm?  
D. A

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Gas Laws Answers Key~~  
The three fundamental gas laws discover the relationship of pressure, temperature, volume and amount of gas.

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Boyle's Law tells us that the volume of gas increases as the pressure decreases. Charles' Law tells us that the volume of gas increases as the temperature increases. And Avogadro's Law tell us that the volume of gas increases as the amount of gas increases. The ideal gas law is the combination of the three simple gas laws.

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