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the other cold, are placed in contact with each other, then the hot body loses heat and becomes colder and the cold body gains heat and becomes hotter, and this process continues till the thermal equilibrium is reached.

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Thermodynamics Problems and

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$\rho gh = P_{\text{pipe}} - P_{\text{atm}}$ and therefore.
 $h = (P_{\text{pipe}} - P_{\text{atm}}) / \rho g = (135000$
 $\text{Pa} - 92000 \text{ Pa}) / (1000 \text{ kg/m}^3 \times$
 $9.81 \text{ m/s}^2) = 4.4 \text{ m.}$ School of
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26) Explain The Laws Of
Thermodynamics.? Zeroth law: If
any two systems are in thermal

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equilibrium with the third system, then they are also in thermal equilibrium with each other. First law: First law of thermodynamic states that energy can neither be created nor be destroyed but it can only be converted from one form to another.

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and Answers. 1. Define
thermodynamic system. A
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as a quantity of matter or a
region in space, on which the
analysis of the problem is
concentrated. 2. Name the ...

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Ideal gas law □ 1. st. form : $pV = nRT$ =
 p □ V □ n □ R □ T □ = Pressure,
 V □ = Volume, n □ = number of

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molecules, k_B = Boltzmann's constant, T = temperature [in K]

2. nd. form : $U = nRT$
 n = number of moles, R =

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Q1: "Heat cannot by itself flow from a body at a lower temperature to a body at a higher temperature" is a statement or consequence of. (a) The second law of thermodynamics. (b) conservation of momentum. (c)

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conservation of mass.

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2019, the exam format is changed so that you need only answer ALL Section A (1-3 Thermodynamics, 4-6 statistical mechanics), ONE question from section B (Thermodynamics) and ONE from section C (Statistical mechanics).

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Thermodynamics

Thermodynamics – Mechanical
Engineering Multiple choice

Questions : 101. Addition of heat
at constant pressure to a gas
results in (a) raising its

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temperature (b) raising its pressure (c) raising its volume (d) raising its temperature and doing external work (e) doing external work. Ans: d. 102. Carnot cycle has maximum efficiency for (a) reversible engine

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aspirants in realizing the question pattern as well as help in analyzing weak & strong areas.

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