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1002 VSEPR Theory

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VSEPR Lab Activity--ANSWER KEY-2 -
CHEM 1A VSEPR Theory ... Species
Name: Lewis Dot Structure: Electronic
Arrangement: Molecular Geometry: BeF
2: linear: linear: BCl 3: trigonal planar:
trigonal planar: CCl 4: tetrahedral

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The valence shell electron pair repulsion
(VSEPR) theory (or “VESPER” for short)
is how the geometry of a molecule is
determined around a central atom. The
molecular geometry main shapes are
tetrahedral, trigonal planar, trigonal
pyramidal, bent, and linear and are named
by measuring the bond angles between the
central atom and another atom bonded to
it.

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0=6*6-3 §=C=:O. 2 linear linear sp

N-×7=-3 μ a tetrahedral

Trpicpgoanmialdae sp suis B.=3 Answer

key - CHEMISTRY The valence shell

electron pair repulsion (VSEPR) theory

(or “VESPER” for short) is how the

geometry of a molecule is determined

around a central atom. The molecular Page

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VSEPR Online Lab Manual In this computer-based problem lab, you predict the structures and shapes of a wide variety of molecules. Although all the work can be done using the online facilities (and some imagination), a variety of molecular models will be made available to assist your visualization.

VSEPR Online Lab Manual - University of Saskatchewan

The VSEPR (Valence Shell Electron Pair Repulsion) model is used to predict the geometry of molecules based on the number of effective electron pairs around a central atom. The main postulate for the VSEPR theory is that the geometrical structure around a given atom is principally determined by minimizing the

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repulsion between effective electron pairs.

17: VSEPR Theory and Shapes of
Molecules (Experiment ...

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lone pairs to the central atom. Then,
compare the model to real molecules!

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VSEPR Theory: Shapes of Molecules -
Part D. When working on VSEPR
experiment: 1. Completely answer all
questions and fill in all blanks. 2. Draw all
Lewis structures. 3. If present, show

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nonbonding electron pairs (or lone pairs) on both central and non-central atoms in Lewis structures. 4.

Chemistry 115 Lab - VSEPR Theory:
Shapes of Molecules

Utilize this visual, active VSEPR Molecular Geometry Balloon introduction lab, in print and digital Google Apps format, illustrating VSEPR 3D molecular shapes. Students begin this activity with a short reading over electron repulsion and draw Lewis structures of 10 molecules and name them.

VSEPR Theory Model Balloon Shapes
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